1903/102 APPLIED SCIENCE AND LABORATORY PRACTICE June/July 2020 Time: 3 hours



THE KENYA NATIONAL EXAMINATIONS COUNCIL

CRAFT CERTIFICATE IN FOOD PROCESSING AND PRESERVATION TECHNOLOGY

MODULE I

APPLIED SCIENCE AND LABORATORY PRACTICE

3 hours

INSTRUCTIONS TO CANDIDATES

You should have the following for this examination:

answer booklet:

non-programmable scientific calculator.

This paper consists of TWO sections; A and B.

Answer ALL the questions in section A and any TWO questions from section B in the answer booklet provided.

 \mathbf{L} h question in section \mathbf{A} carries $\mathbf{4}$ marks while each question in section \mathbf{B} carries $\mathbf{20}$ marks.

Maximum marks for each part of a question are as shown.

Candidates should answer the questions in English.

This paper consists of 4 printed pages.

Candidates should check the question paper to ascertain that all the pages are printed as indicated and that no questions are missing.

SECTION A (60 marks)

Answer ALL questions in this section.

- 1. Differentiate between osmosis and active transport. (4 marks)
- 2. Explain the effect of water on the rate of growth of micro-organisms. (4 marks)
- 3. The Figure 1 shows a side view of the human male reproductive system.
 - (a) Name the parts labelled A and B. (2 marks)
 - (b) Explain the functions of the part labelled C. (2 marks)

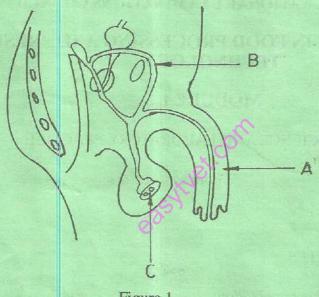


Figure 1

- 4. List four methods of laboratory waste disposal. (4 marks)
- 5. Explain each of the following as used in glass blowing technology:
 - (a) fire polishing; (2 marks)
 - (b) glass bending. (2 marks)
- 6. State four precautions taken when storing cryogens. (4 marks)
- 7. Name four utilities that should be incorporated when renovating a building into a laboratory. (4 marks)
- 8. State four symptoms shown by a person affected by electric shock. (4 marks)

9.		ement X, consists of three isotopes with mass number 22,24 and 25 with percentance of 89.6%, 6.4% and 4.0% respectively. Calculate the relative atomic master X.	
10.	Disting	guish between each of the following:	
	(a)	atomic mass and mass number;	(2 marks)
	(b)	element and compound.	(2 marks)
11.		mensions of a cuboid are 6.35 cm by 4.75 cm by 0.74 cm. Calculate its volun the answer in standard form.	ne (4 marks)
12.	Descri	be each of the following types of fluid flow:	
	(a)	Laminar flow;	(2 marks)
0	(b)	Turbulent flow.	(2 marks)
13.	With t	he aid of a diagram, explain the quarter sawing method of timber conversion.	(4 marks)
14.	Explai	n the functions of each of the following parts of vernier calipers:	
	(a)	Vernier scale;	(2 marks)
	(b)	Main scale.	(2 marks)
15.	Name	four tools used in technical drawing.	(4 marks)
		SECTION B (40 marks)	
0		Answer any TWO questions from this section.	
16.	(a)	Define each of the following ecological terms:	
		(i) Population;	(2 marks)
		(ii) Community.	(2 marks)
	(b)	State six important functions of water in living organisms.	(6 marks)
	(c)	Explain five effects of man's activities on the environment.	(10 marks)

17.	(a)	Name five personal protective equipment for a laboratory technician.	(5 marks)
	(b)	State fifteen safe storage methods of chemicals in the laboratory.	(15 marks)
18.	(a)	State six uses of halogens and their compounds.	(6 marks)
	(b)	Write the chemical equations for the reactions between:	
		(i) Magnesium and dilute sulphuric acid;(ii) Aluminium and dilute hydrochloric acid.	(2 marks) (2 marks)
	(c)	Water with negligible viscosity flows steadily through a horizontal pipe of vacross-section area. At point A of cross section area 10 cm ² , the velocity is 0. Calculate the:	100
		(i) velocity at point B of cross section area 2.5 cm ² .	(4 marks)
		(ii) pressure difference between points A and B, given that the density of water is 1000 kg/m ³ .	(6 marks)
19.	(a)	Difference between front elevation and end elevation as used in technical dra	awing. (4 marks)
	(b)	State four characteristics of hardwood trees.	(4 marks)
	(c)	Explain the safe storage of timber in a closed building.	(12 marks)

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